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2-Amino-1,1,3-tricyanopropene: a new inhibitor of oxygen evolution in photosynthesis

One of the most fundamental problems of photosynthesis is the mechanism of O_2 evolution; despite many studies of this problem, remarkably little is known. Investigations by Wessels and Van Der Veen¹ showed that some substituted phenylureas (e.g. DCMU) act as inhibitors of O_2 evolution; further work (Gingras and Lemasson², Izawa and Good³, has been carried out with these inhibitors on the mechanism of inhibition which may throw light upon the O_2 evolution process itself. Some substituted benzimidazoles and aminotriazines also inhibit the oxygen-evolving system of photosynthesis (Büchel et al.⁴, Bishop⁵). The purpose of the present work was to search for other compounds capable of specifically inhibiting O_2 evolution. 2-Amino-I,I,3-tricyanopropene (TCAP), previously shown to uncouple phosphorylation in mitochondria (Eberts⁶, Parker²), was found in this study to inhibit selectively O_2 evolution of the photosynthetic reactions studied (Arnon et al.⁶).

Fresh spinach leaves obtained commercially were used in the preparation of chloroplasts by a modification of the method of Whatley and Arnon⁹. 50 g of sliced,

Abbreviations: TCAP, 2-amino-1,1,3-tricyanopropene; DCMU, 3-(3,4-dichlorophenyl)-1,1-dimethylurea; DCIP, 2,6-dichlorophenolindophenol; PMS, phenazine methosulphate.

deribbed leaf blades were ground with 100 ml of sucrose medium (0.4 M sucrose, 0.02 M Tris-HCl, pH 8.0, 0.01 M NaCl) and 50 g of cold sand. The green juice obtained by squeezing the slurry through cheesecloth was centrifuged for I min at $200 \times g$ to sediment sand, leaf debris and whole cells. The green supernatant liquid was decanted and centrifuged for 7 min at $4000 \times g$. As a result of this centrifugation most of the intact chloroplasts were sedimented: they were resuspended in 50 ml of the sucrose grinding medium and again centrifuged for 7 min at 4000 \times g. 10 ml of 0.05 M Tris-HCl (pH 8.0) were then added to the sedimented chloroplasts to yield a suspension of broken chloroplasts. All experiments were carried out in 14-ml Warburg flasks using a Braun photosynthetic Warburg bath (TypeV185). Reactions were started by tipping the ³²P and oxidant (where required). Illumination was 1000 ft-candles and O₂ evolution was measured manometrically. ATP was determined as organic ³²P by the method of Hagihara and Lardy¹⁰ and an aliquot counted in a Geiger counter. TPNH and ferricyanide were measured spectrophotometrically at 340 m μ and 400 m μ respectively. Dark controls were measured in all cases. Ferredoxin was prepared by the method of Tagawa and Arnon¹¹. TCAP was kindly supplied by Upjohn Ltd., Crawley, England. Chlorophyll was estimated by the method of Arnon¹².

TCAP, at low concentrations, inhibits O_2 evolution in all non-cyclic systems tested where the evolution of O_2 is an integral step of the electron-transport chain (e.g. for 50% inhibition of TPN+-H₂O, [TCAP] = $1.3 \cdot 10^{-5}$ M; ferricyanide-H₂O, 8.0 · 10^{-5} M; and ferricyanide-DCIP, $1.1 \cdot 10^{-4}$ M) (see also Table I). In the only non-cyclic system tested where electron transport does not involve the evolution of O_2 , namely, TPN+-ascorbate plus DCMU, the photoreduction of TPN+ and the accompanying photophosphorylation were both unaffected (Table I). During these

TABLE I EFFECT of TCAP on O_2 evolution and on phosphorylation

The reaction mixture included in a final volume of 3 ml, chloroplast fragments containing 0.2 mg chlorophyll, and the following (in μ moles): Tris–HCl buffer (pH 8.0), 75; MgSO₄, 5; ADP, 10; $K_2H^{32}PO_4$, 10; and the following additions (in μ moles): in Expt. A: $K_3Fe(CN)_6$, 15; Expt. B: $K_3Fe(CN)_6$, 15; DCIP, 0.2; Expt. C: TPN+, 4 (plus spinach ferredoxin); Expt. D: TPN+, 4 (plus spinach ferredoxin); sodium ascorbate, 20; DCIP, 0.2; Expt. E: PMS, 0.1; Expt. F: Vitamin K_3 , 0.3; sodium ascorbate, 10; DCMU, 0.3 (see refs. 13, 14); Expt. G: Vitamin K_3 , 0.3; sodium ascorbate, 10. All reactions were carried out at 15° under argon except Expt. G, which was under air. Light intensity was 1000 ft-candles. TCAP was dissolved in 50% methanol in water.

Expt.	Percentage inhibition by $TCAP$ (2·10 ⁻⁴ M)	
	Electron transport	
Non-cyclic photophosphorylation		
A. Ferricyanide-H ₂ O	83	81
B. Ferricyanide-DCIP	66	
C. TPN+-H ₂ O	100	100
D. TPN+-ascorbate	О	О
Cyclic and pseudocyclic photophosphorylation		
E. PMS		О
F. Vitamin K ₃ , anaerobic		26
G. Vitamin K ₂ , aerobic	-	88

studies, it was found that ascorbate reacted with TCAP in the presence of chloroplasts and light to form a substance having significant light-absorbing properties at 340 mu: therefore appropriate controls were included in all experiments involving the TPN+ascorbate system thus permitting correction of the TPNH assays. At considerably higher concentrations, TCAP inhibits anaerobic cyclic photophosphorylation, catalysed by vitamin K_3 (8·10⁻⁴ M TCAP for 50 % inhibition) and at even higher concentrations, photophosphorylation catalysed by PMS (2·10⁻³ M TCAP for 50 % inhibition).

The experimental results (Table I) show that TCAP is a potent inhibitor of O₂ evolution and consequently of any of the reactions of photosynthesis dependent upon water as a source of electrons. In this respect, TCAP resembles the action of the substituted phenylureas which also curtail O₂ evolution at very low concentrations. The similarities of the effects of these inhibitors extend to cyclic photophosphorylation where inhibition occurs at higher concentration of inhibitors; however, the chemical structures of these inhibitors show few likenesses.

TCAP is much more effective as an inhibitor of O₂ evolution in photosynthesis than as an uncoupler of oxidative phosphorylation where a 50 % inhibition demands a concentration of inhibitor of 8·10⁻⁴ M which is approximately equivalent to the concentration required for a similar inhibition of anaerobic cyclic photophosphorylation catalysed by vitamin K₃.

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